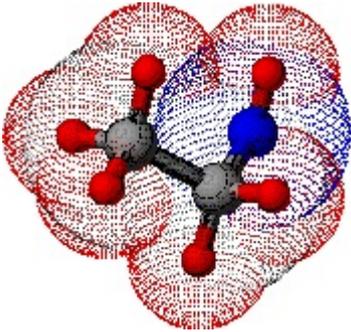
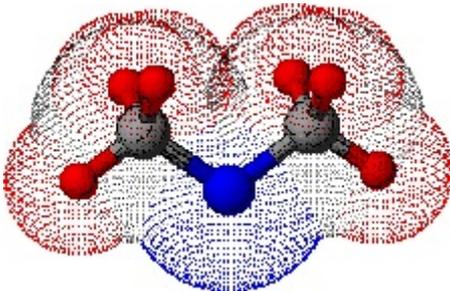
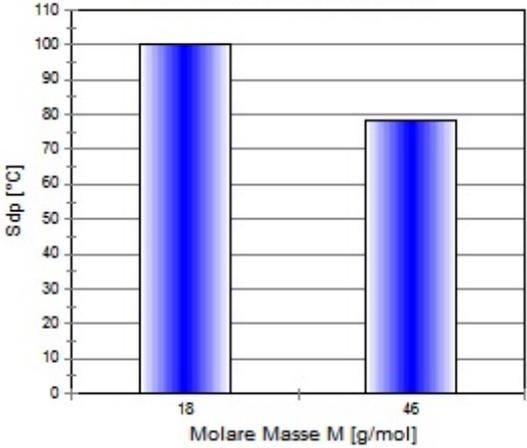
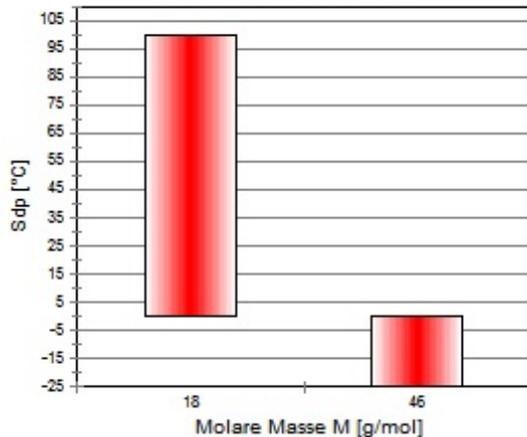


C₂H₆O - eine Summenformel, zwei Isomere

Ethanol	Dimethylether																		
																			
$\begin{array}{c} \text{H}_3\text{C} - \text{C} - \text{OH} \\ \\ \text{H}_2 \end{array}$	$\begin{array}{c} \text{H}_3\text{C} - \text{O} - \text{CH}_3 \\ \\ \text{O} \end{array}$																		
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Arbeitsauftrag:

Vergleiche **Ethanol** mit **Dimethylether**, stelle Gemeinsamkeiten und Unterschiede fest und begründe diese **schriftlich**!